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The interplay of these factors will determine the degree of success in accomplishing either disinfection or sterilization. in all situations, review the manufacturer's recommendations for correct formulation and use. do not attempt to use a chemical disinfectant for a purpose it was not designed for.1 hugo et al. principles and practice of disinfection, preservation and sterilization. third edition, blackwell science. 1999, pg. 54. chemical sterilization with peracetic acid and ethylene oxide sterilization for long narrow lumens. am. j. infect. control 1998; 26:469-77. Chapter 13 - sterilization & disinfection 297 chapter 13 - lesson 1 sterilization introduction sterilization is the process of killing all microorgan-isms (bacterial, viral, and fungal) with the use of ei-ther physical or chemical agents. a disinfectant is a chemical substance that kills microorganisms on in-Tween® 80 product number p 4780 store at room temperature product description cas number: 9005-65-6 specific gravity: 1.07 (25 °c) hlb (hydrophile-lipophile balance) value: 15.01,2 critical micellar concentration (cmc): 13-15 mg/liter 1,3 brookfield viscosity: 400-620 cps (25 °c, neat) Principles of sterilization and disinfection. in order to implement a laboratory biosafety program it is important to understand . the principles of decontamination, cleaning, sterilization, and disinfection. we review here the definitions of sterilization, disinfection, antiseptics, decontamination, and sanitization to avoid misuse and confusion.